

Sample Preparation for MALDI-TOF

Samples for MALDI-TOF analysis need to meet certain requirements for obtaining good spectra. The more careful you prepare samples (including early steps of isolation and preparation) the more likely a successful analysis will be. Here are some guidelines of which kind of treatment is advantageous for mass spectrometric analysis and which is not:

- ▶ Avoid the use of non-volatile agents like salts (NaCl, CaCl₂, KH₂PO₄), detergents (Tween, Triton, SDS), chaotropic agents (Urea, Guanidinium salts) and solvents like DMSO or Glycerol.

- ▶ If you can't avoid these agents, purify. Dialysis and RP-HPLC are good purification methods if you use volatile solvents and buffers. After purification, lyophilize if possible. Ion exchange beads may work well for salt removal.

- ▶ Suitable solvents are ones that are volatile. For sample work up and purification: water, ammonium hydrocarbonate, ammonium acetate, ammonium formiate, acetonitrile, trifluoroacetic acid.

- ▶ Quantitate the sample you are going to provide for analysis by methods like: photometry (OD), and ELISA. HPLC is useful since it allows for purification and quantitation in a single procedure. The range for many samples/preparations is not very large,

therefore it is necessary to have a good estimate of the sample amount because the sample amount may need to be varied on the target.

- ▶ The total amount of sample needed for MALDI analysis depends on the sample type.

- ▶ Give information like: structure, sequence, molecular weight, type of compound, biological activity, chemical reactivity, pH, sample amount/concentration, describe purification/isolation with focus on relative agents/solvents, known or suspected impurities, suitable solvents, hazardous properties: radioactivity, carcinogenicity, poison, or explosive.

For removing contaminants and interfering substances you can use : Microcon centrifugal filtration (3 kDa cut-off), Mini-dialysis, TCA acetone precipitation, C18 or C4 micro-reversed-phase chromatography (Zip tipping), and Amersham 2D clean-up. When submitting samples, please use a sample container of appropriate size. Two μL of sample is more easily recovered from 0.5mL Eppendorf tubes than from 1.5mL . Prerinsing the tube with methanol or acetonitrile lowers the chemical background for low concentration samples or complex mixtures.

Table shows the *MAXIMUM* concentration of surfactants, buffers and salts permissible in MALDI. *N/A* means that technique is not compatible with *ANY* amount of indicated compound. (Modified from “The Expanding Role of Mass Spectrometry in Biotechnology” by Gary Siuzdak, 2003, p. 84-85)

Surfactants, Buffers, Salts	Concentration, mM
<i>TRIS</i>	100
<i>HEPES</i>	100
<i>BICINE</i>	50
<i>Urea</i>	500
<i>Guanidine, HCl</i>	250
<i>Dithiothreitol</i>	500
<i>Glycerol</i>	130
<i>n-Octyl-β-glucopyranoside</i>	3.4
<i>n-Octyl-sucrose</i>	N/A
<i>n-Dodecyl sucrose</i>	N/A
<i>n-Dodecyl maltoside</i>	N/A
<i>Octyl thioglucoside</i>	N/A
<i>n-Hexyl glucoside</i>	N/A
<i>n-Dodecyl glucoside</i>	N/A
<i>PEG1000</i>	N/A
<i>PEG2000</i>	0.5
<i>Triton X-100</i>	1.6
<i>NP-40</i>	1.7
<i>Zwittergent, 3-16</i>	2.6
<i>Tween20</i>	N/A
<i>Thesit</i>	N/A
<i>SDS</i>	0.35
<i>LDAO</i>	4.4
<i>CTAB</i>	N/A
<i>CHAPS</i>	0.16
<i>Sodium Cholate</i>	N/A
<i>Sodium Taurocholate</i>	N/A
<i>Sodium Azide</i>	15
<i>NH₄CHO₃</i>	50
<i>NaCl</i>	50
<i>Sodium Acetate</i>	50
<i>NaHPO₄</i>	10
<i>TFA</i>	N/A